

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet



Synthesis of poly-substituted tetrahydropyridines from Baylis–Hillman adducts modified with *N*-allylamino group via radical cyclization

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ARTICLE INFO

Article history: Received 19 February 2009 Accepted 27 February 2009 Available online 6 March 2009

Keywords: Radical cyclization Tetrahydropyridines Baylis-Hillman adducts

ABSTRACT

An expedient method was developed for the synthesis of 1,4,5,6-tetrahydropyridines by radical cyclization protocol involving consecutive 1,5-hydrogen transfer and double bond isomerization process starting from Baylis–Hillman adducts.

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Radical cyclizations have been used for the synthesis of various cyclic compounds. In some instances initial radical species underwent 1,*n*-H transfer to form another radical before cyclization reaction. ^{1,2} Among the 1,*n*-H transfers, 1,5- and 1,6-H transfers are the most common. ^{1,2} Very recently we observed an interesting radical cyclization procedure for the synthesis of tricyclic lactam derivatives involving 1,5-H transfer and concomitant isomerization. ^{1a}

Suitably substituted dihydro- and tetrahydropyridine derivatives have been regarded as important synthetic intermediates for the synthesis of various important compounds.³ Especially, 1,4,5,6-tetrahydropyridine derivative **A** has been used for the synthesis of antidepressant drug paroxetine (**B**)⁴ and many paroxetine-like PSSRIs (phenylpiperidine selective Serotonin reuptake inhibitors) including femoxetine (**B**').^{4,5} In addition, many 1,2,5,6-tetrahydropyridine derivatives **C** have been reported as renin inhibitors (Fig. 1).⁵ During our recent studies on the chemical transformations of Baylis–Hillman adducts,^{6,7} we imagined that we could synthesize tetrahydropyridine skeleton^{4,5} via the radical cyclization of **4a-e** as in Scheme 1. *N*-Tosyl-*N*-allyl derivatives **4a-e** were prepared in good to moderate yields from the acetates of Baylis–Hillman adducts **1a** and **1b** as summarized in Scheme 2 in two steps.^{7,8}

With the substrates $\mathbf{4a-e}$ we examined the radical cyclization reaction under the conditions of $n\text{-Bu}_3\text{SnH}$ (1.2 equiv)/AIBN in refluxing benzene.⁷ As expected, we obtained 1,4,5,6-tetrahydropyridines $\mathbf{5a-e}$ in good to moderate yields (56–82%) in short time (Table 1).⁹ Compounds having two stereogenic centers including $\mathbf{5a}$, $\mathbf{5c}$, and $\mathbf{5d}$ were isolated as their syn/anti diastereomeric mix-

ture in a ratio of 3:2–4:1. However, we could not separate each isomer in their pure form. The mechanism for the formation of **5** could be regarded as in Scheme 1: (i) 1,5-hydrogen transfer from the initial radical (**II**) to form (**II**), ^{1a} isomerization to more stable benzylic radical (**III**), the following cyclization in a 6-exo-trig mode to (**IV**), and final hydrogen radical abstraction to **5**.

However, the reaction of N-benzyl derivative 4f showed low yield of product **5f** (38%) under the same conditions (1.2 equiv of *n*-Bu₃SnH) due to the formation of many intractable side products. Fortunately, we obtained **5f** in an improved yield (76%) when we used n-Bu₃SnH in excess amounts (2.5 equiv) presumably due to rapid hydrogen abstraction of the corresponding radical intermediate (IV) from *n*-Bu₃SnH (Scheme 3). Similarly, the reaction of *N*phenyl derivative 4g showed similar pattern. When we used 1.2-1.5 equiv of n-Bu₃SnH, pyrrolidine derivative **6** was formed in appreciable amounts with low yield of 5g. The desired compound **5g** was isolated in moderate yield (50%) with 2.5 equiv of n-Bu₃SnH, together with pyrrolidine derivative 6 in 21% yield as a syn/anti (1:1) mixture (Scheme 4). The formation of 6 could be explained as sequential hydrostannylation at the allyl group 10 and following radical cyclization in a 5-exo-trig mode. The reduction of bromophenyl moiety might occur independently with excess Bu₃SnH. Compound **6** was also prepared from **4g** under the same conditions (2.5 equiv of *n*-Bu₃SnH) in good yield (77%). We obtained compounds 5h (31%) and 7 (24%) from the reaction of N-methallyl derivative 4h, similarly. In the reaction, two minor compounds 8 (12%) and 9 (28%) were isolated together as side products and we did not confirm the geometry of double bond Scheme 5.

In summary, we disclosed an efficient synthetic way for 1,4,5,6-tetrahydropyridines by radical cyclization protocol involving

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Figure 1.

Scheme 1.

Table 1Synthesis of tetrahydropyridine derivatives

Entry	Conditions ^a	Substrate 4 (%)	Conditions ^b (h)	Product 5 (%)
1	2a + 3a ,6 h	4a (80)	2	5a $(72, syn/anti = 2:3)^c$
2	2a + 3b , 4 h	4b (93)	1	5b (70)
3	2a + 3c , 3 h	4c (80)	2	5c (82, syn/anti = 2:3) ^c
4	2a + 3d , 24 h	4d (87)	2	5d (80, $syn/anti = 1:4$) ^c
5	2b + 3b , 4 h	4e (92)	3	5e (56)

- $^{\rm a}$ Conditions: Compound **2** (1.0 mmol), compound **3** (1.5 mmol), K₂CO₃ (1.2 equiv), DMF, rt.
- b Conditions: Substrate **4** (0.5 mmol), *n*-Bu₃SnH (0.6 mmol), AlBN (cat) benzene, reflux.
- ^c The ratio of *syn/anti* was determined in ¹H NMR and is arbitrary.

OAC

TSNH₂ (2.0 equiv)

$$K_2CO_3$$
 (2.0 equiv)

DMF, 50 °C, 3 h

2a: EWG = COOMe (66%)
2b: EWG = COOEt (62%)

 R_1
 K_2CO_3 (1.2 equiv)

DMF, R_3
 R_1
 R_2
 R_3
 R_1
 R_2
 R_3
 R_1
 R_2
 R_3
 R_1
 R_3
 R_1
 R_3
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 R_3
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 R_3
 R_4
 R_3
 R_4
 R

Scheme 2.

Scheme 3.

Scheme 4.

Scheme 5.

consecutive 1,5-hydrogen transfer and double bond isomerization process. Applications of this methodology are currently underway for the synthesis of paroxetine derivatives having 5-alkyls.

Acknowledgments

This work was supported by the Korea Research Foundation Grant funded by the Korean Government (MOEHRD, KRF-2008-313-C00487). Spectroscopic data were obtained from the Korea Basic Science Institute, Gwangju branch.

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- 9. Typical experimental procedure for the synthesis of **4b** and **5b**: To a stirred mixture of **2a** (424 mg, 1.0 mmol) and **3b** (203 mg, 1.5 mmol) in DMF (3 mL) was added K₂CO₃ (166 mg, 1.2 mmol) and stirred at room temperature for 4 h. After the usual aqueous workup and column chromatographic purification process (hexanes/ether, 2:1) compound **4b** was isolated as colorless oil, 445 mg (93%). A mixture of **4b** (239 mg, 0.5 mmol), Bu₃SnH (175 mg, 0.6 mmol), and AlBN (8 mg, 0.05 mmol) in benzene (3 mL) was heated to reflux for 1 h. After removal of solvent, the residue was purified by column chromatography (hexanes/ether, 4:1) to obtain compound **5b** as a white solid, 279 mg (70%). Selected spectroscopic data of compound **4b**, **5b**, **5f**, and **6** are as follows.Compound **4b**: 93%; colorless oil; IR (film) 1720, 1436, 1339, 1248, 1159 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 1.62 (s, 3H), 2.41 (s, 3H), 3.65 (s, 3H), 3.71 (s, 2H), 4.15 (s, 2H), 4.77 (s, 1H), 4.78 (s, 1H), 7.20–7.27 (m, 3H), 7.37 (t, *J* = 7.5 Hz, 1H), 7.46 (d, *J* = 7.8 Hz, 1H), 7.56–7.65 (m, 3H), 7.76 (s, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 19.93, 21.46, 44.09, 52.06, 54.89, 113.58, 124.11, 127.36, 127.38, 129.40, 129.55, 130.33, 130.98, 132.81, 134.63, 136.59, 140.82, 142.32, 143.00, 167.29; ESIMS m/z 478 (M*+1), 480 (M*+3).Compound **5b**: 70%; white

Lett. 1999, 659-660.

solid, mp 110–112 °C; IR (KBr) 1709, 1633, 1168, 1096 cm $^{-1};$ ¹H NMR (CDCl₃, 300 MHz) δ 0.59 (s, 3H), 0.88 (s, 3H), 2.49 (s, 3H), 2.80 (d, J = 12.0 Hz, 1H), 3.18 (d, J = 12.0 Hz, 1H), 3.37 (s, 1H), 3.61 (s, 3H), 6.70 (d, J = 5.7 Hz, 2H), 7.03–7.15 (m, 3H), 7.40 (d, J = 8.1 Hz, 2H), 7.79 (d, J = 8.1 Hz, 2H), 8.16 (s, 1H); $^{13}\mathrm{C}$ NMR (CDCl₃, 75 MHz) δ 21.64, 25.25, 27.03, 31.96, 48.00, 49.38, 51.43, 110.10, 126.62, 127.33, 127.76, 128.66, 130.09, 134.07, 134.44, 141.23, 144.69, 167.04; ESIMS m/z 400 (M*+1). Anal. Calcd for $C_{22}H_{25}\mathrm{NO_4S}$: C, 66.14; H, 6.31; N, 3.51. Found: C, 66.37; H, 6.13; N, 3.45.Compound 5f: 76%; white solid, mp 123–24 °C; IR (KBr) 1682, 1620, 1167, 1144 cm $^{-1};$ ¹H NMR (CDCl₃, 300 MHz) δ 0.54 (s, 3H), 0.99 (s, 3H), 2.36 (d, J = 12.3 Hz, 1H), 2.85 (d, J = 12.3 Hz, 1H), 3.35 (s, 1H), 3.56 (s, 3H), 4.37 (d, J = 15.3 Hz, 1H), 4.44 (d, J = 15.3 Hz, 1H), 7.02 (d, J = 6.6 Hz, 2H), 7.10–7.41 (m, 8H), 7.77 (s, 1H); $^{13}\mathrm{C}$ NMR (CDCl₃, 75 MHz) δ 25.51, 28.07, 31.52, 47.85, 50.51, 52.41, 60.11, 97.51, 125.91, 127.46, 127.90, 127.98, 128.71, 128.78, 136.59, 144.32, 144.55, 168.89; ESIMS m/z 336 (M*+1). Anal. Calcd for $C_{22}H_{25}\mathrm{NO}_2$: C, 78.77; H, 7.51; N, 4.18. Found: C, 78.89; H, 7.76;

N, 4.05.Compound **6** (separated pure isomer, but syn/anti was not determined): 21%; colorless oil; IR (film) 1726, 1598, 1507, 1374 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 0.85–0.92 (m, 15H), 1.02–1.07 (m, 2H), 1.20 (t, J = 7.2 Hz, 3H), 1.24–1.55 (m, 12H), 2.46–2.53 (m, 1H), 2.74 (d, J = 13.8 Hz, 1H), 2.94 (d, J = 8.7 and 6.3 Hz, 1H), 3.34 (d, J = 13.8 Hz, 1H), 3.18 (d, J = 9.9 Hz, 1H), 3.56 (d, J = 9.9 Hz, 1H), 3.56-3.62 (m, 1H), 4.09–4.20 (m, 2H), 6.50 (d, J = 7.5 Hz, 2H), 6.66 (t, J = 7.5 Hz, 1H), 7.12–7.29 (m, 7H); ¹³C NMR (CDCl₃, 75 MHz) δ 9.20, 9.31, 13.69, 14.30, 27.41, 29.19, 41.24, 45.70, 51.87, 54.42, 59.34, 60.44, 111.17, 115.52, 126.66, 128.26, 129.12, 129.92, 137.67, 147.39, 173.31; ESIMS m/z 613 (M $^+$ +1). 10. For the similar hydrostannylation of double bond and the following radical cyclizations, see: (a) Gerbino, D. C.; Koll, L. C.; Mandolesi, S. D.; Podesta, J. C. Organometallics **2008**, 27, 660–665; (b) Hanessian, S.; Leger, R. J. Am. Chem. Soc. **1992**, 114, 3115–3117; (c) Miura, K.; Saito, H.; Uchinokura, S.; Hosomi, A. Chem.